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Mr. Bastien - Home

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Math Handbook Transparency Masters Chemistry: Matter and Change 93 Addition and Subtraction Before numbers in scientific notation can be added or subtracted, the exponents must be equal. (3.4 ! 10²) " (4.57 ! 10³) # (0.34 ! 10³) " (4.57 ! 10³) Not equal Equal The decimal is moved to the left to increase the exponent. # (0.34 " 4.57) ! 10³ # 4.91 ! 10³ Multiplication

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6 Math Handbook Transparency Worksheet 2 - Afreshcutflorist

92 Chemistry: Matter and Change Math Handbook Transparency Worksheets 1. Express each of the following numbers in scientific notation. a. 230 b. 5601 c. 14 100 000 d. 56 million e. $\frac{2}{10}$ f. 0.450 13 g. 0.089 h. 0.000 26 i. 0.000 000 698 j. 12 thousandth 2. Express each of the following measurements in scientific notation.

Scientific Notation Use with Appendix B, Scientific Notation

Perform the following operations and express the answers in scientific notation. a. $(6.0 \times 10^1) \times 10^5$ b. $(7 \times 10^3) \times (1.2 \times 10^4)$ c. $(1.8 \times 10^{-4}) \div (1.2 \times 10^{-2})$ 94 Chemistry: Matter and Change Math Handbook Transparency Worksheets ...

SCIENTIFIC NOTATION Name Example 1: Convert 1,500,000 to ...

96 Chemistry: Matter and Change Math Handbook Transparency Worksheets 1. For each of the measurements in the table below, determine if the underlined number is significant or not significant. Place a check mark in the appropriate box and in the box under the rule that you used to make your determination. 2.

SIGNIFICANT FIGURES Name

The answer is expressed as 190,000 cm since 19 cm has only two significant figures. When adding and subtracting, limit and round your answer to the least number of decimal ... Math Handbook Transparency Worksheets m. 7.01 m/s Chemistry: Matter and Change 96 . Created Date:

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Chemistry Matter And Change Chapter 10 Math Skills ...

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